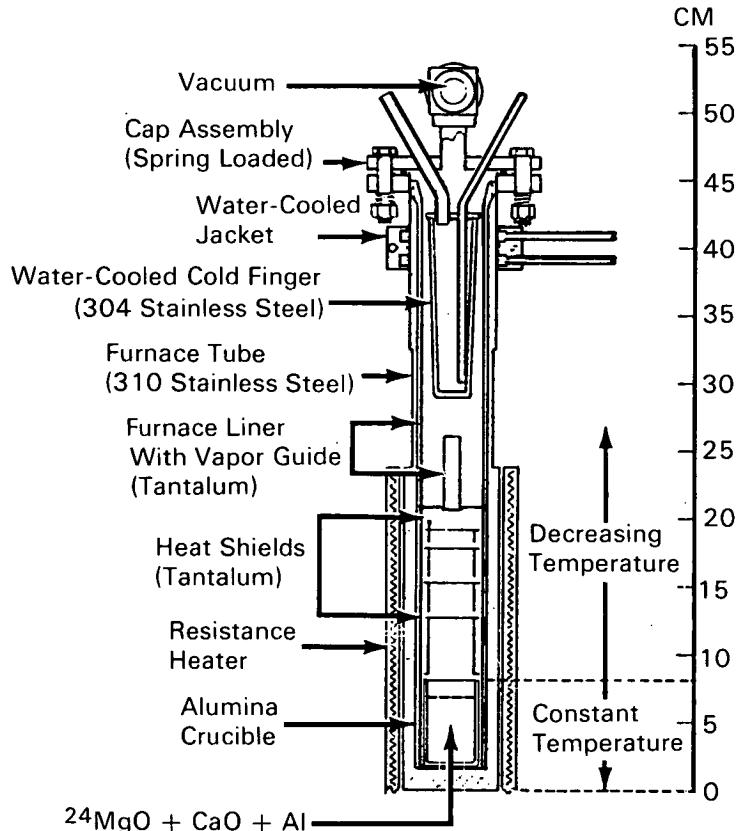


# AEC-NASA TECH BRIEF



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## Isotopically Pure $^{24}\text{Mg}$ Is Prepared From $^{24}\text{MgO}$



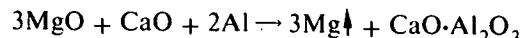
### The problem:

To prepare isotopically pure  $^{24}\text{Mg}$ , suitable for use in neutron scattering and polarization experiments. Because of the value of the pure  $^{24}\text{MgO}$  starting material, the method used must assure a high product yield.

### The solution:

An apparatus (see diagram) which permits thermal reduction of  $^{24}\text{MgO}$  with aluminum and  $\text{CaO}$ , and

subsequent vaporization of the product metal in vacuum according to the equation:



### How it's done:

The apparatus used for the  $^{24}\text{MgO}$  reduction consists of a resistance-heated furnace tube and cap assembly. The tube contains an aluminum oxide crucible for holding the feed charges, tantalum radiation

(continued overleaf)

shields, a tantalum liner with a vapor guide, and a water-cooled cold finger for collection of the vaporized magnesium.

The charges to the reduction furnace are in the form of pellets prepared from a stoichiometric mixture of  $^{24}\text{MgO}$  and  $\text{CaO}$  together with aluminum in excess of the amount necessary. The pellets are outgassed at  $5 \times 10^{-4}$  torr at a temperature between  $600^\circ$  and  $700^\circ\text{C}$ , then heated to  $1150^\circ\text{C}$  and held at this temperature for 2 hours to effect  $^{24}\text{MgO}$  reduction and product distillation.

**Notes:**

1. Product yields of  $^{24}\text{Mg}$  are reported for 7 runs to be 98.3% (std. dev.  $\pm 1.2\%$ ). Mass spectrometric analyses of product samples showed that the at% of magnesium isotopes to be:  $^{24}\text{Mg}$ ,  $99.91 \pm 0.01$ ;  $^{25}\text{Mg}$ ,  $0.06 \pm 0.01$ ;  $^{26}\text{Mg}$ ,  $0.03 \pm 0.01$ .
2. Additional details are contained in "Preparation of  $^{24}\text{Mg}$  from  $^{24}\text{MgO}$ ," by N. R. Chellew, R. K. Steunenberg, and J. D. Schilb, Argonne National Laboratory, which was published in *Nuclear Instruments and Methods*, 44 (1966), p. 149-150.

3. Inquiries concerning this innovation may be directed to:

Office of Industrial Cooperation  
Argonne National Laboratory  
9700 South Cass Avenue  
Argonne, Illinois 60439  
Reference: B68-10293

Source: N. R. Chellew,  
R. K. Steunenberg, and J. D. Schilb  
Chemical Engineering Division  
(ARG-10154)

**Patent status:**

Inquiries about obtaining rights for commercial use of this innovation may be made to:

Mr. George H. Lee, Chief  
Chicago Patent Group  
U.S. Atomic Energy Commission  
Chicago Operations Office  
9800 South Cass Avenue  
Argonne, Illinois 60439